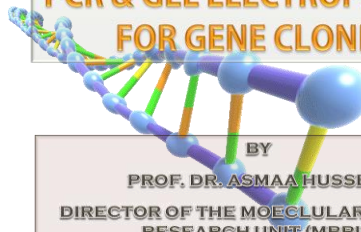


PCR & GEL ELECTROPHORESIS FOR GENE CLONING



BY


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What is PCR?

PCR is an exponentially progressing synthesis of the defined target DNA sequences *in vitro*

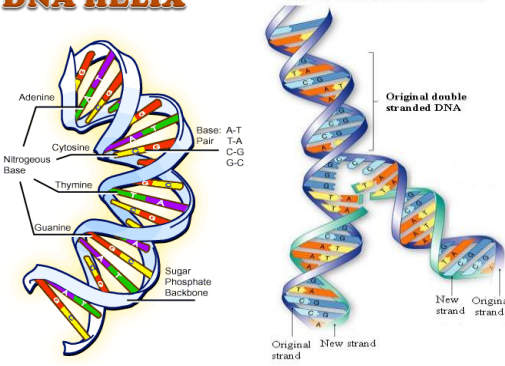
It was invented in 1983 by **Dr. Kary Mullis**, for which he received the Nobel Prize in Chemistry in 1993

POLYMERASE CHAIN REACTION



DNA HELIX

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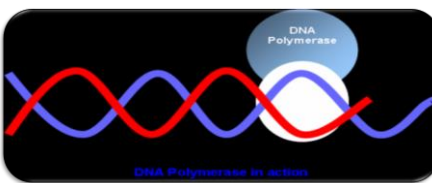


Adenine
Cytosine
Nitrogenous Base
Thymine
Guanine
Base: A-T
T-A
C-G
G-C
Sugar
Phosphate
Backbone
Original double stranded DNA
Original strand
New strand

What is PCR?

Why “Polymerase”?

It is called “polymerase” because the only enzyme used in this reaction is DNA polymerase



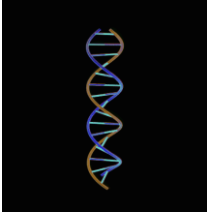

DNA Polymerase

DNA Polymerase in action

What is PCR?

Why "Chain"?

IT IS CALLED "CHAIN" BECAUSE THE PRODUCTS OF THE FIRST REACTION BECOME SUBSTRATES OF THE FOLLOWING ONE, AND SO ON

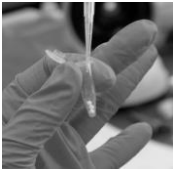

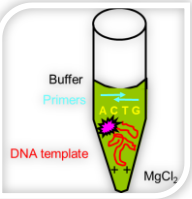


What is PCR?

THE "REACTION" COMPONENTS

- 1) Target DNA → Contains the sequence to be amplified
- 2) Pair of Primers → Oligonucleotides that define the sequence to be amplified
- 3) dNTPs → Deoxynucleotidetriphosphates: DNA building blocks
- 4) Thermostable DNA Polymerase → Enzyme that catalyzes the reaction
- 5) Mg^{++} ions → Cofactor of the enzyme
- 6) Buffer solution → Maintains pH & ionic strength of the reaction solution suitable for the activity of the enzyme

What is PCR?




3) dNTPs



4) Thermostable DNA Polymerase

5) Mg^{++} ions


6) Buffer solution



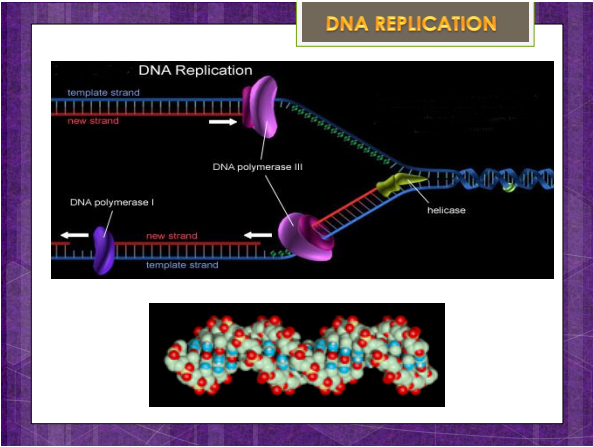
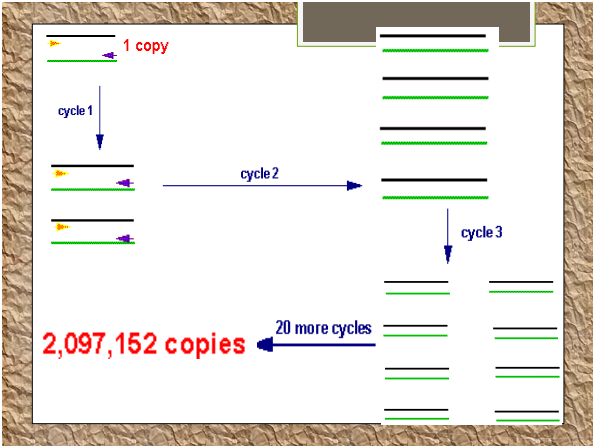
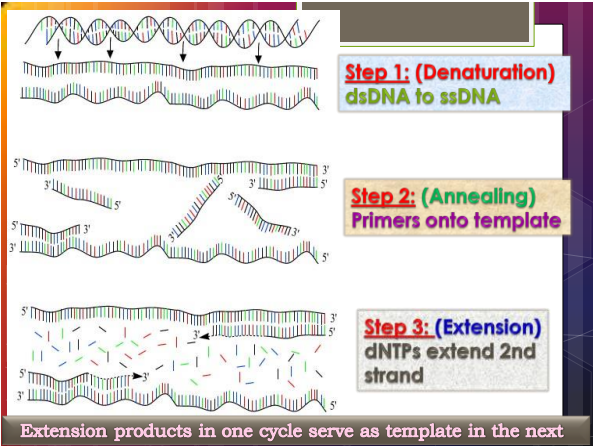
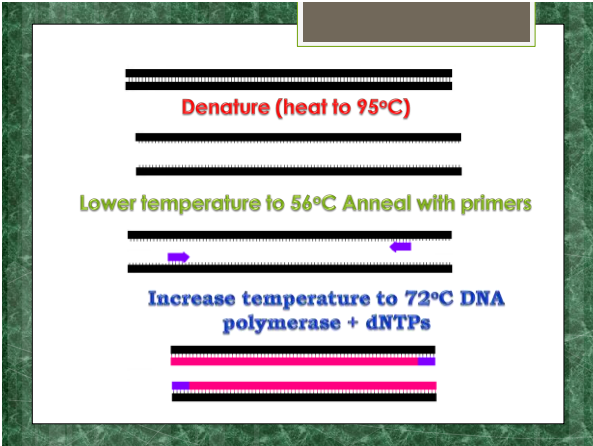
THE REACTION

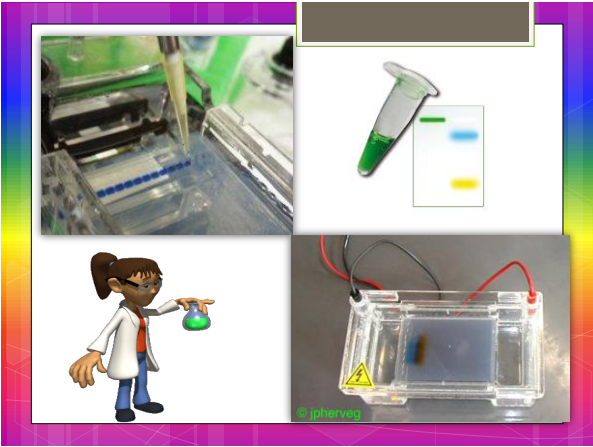
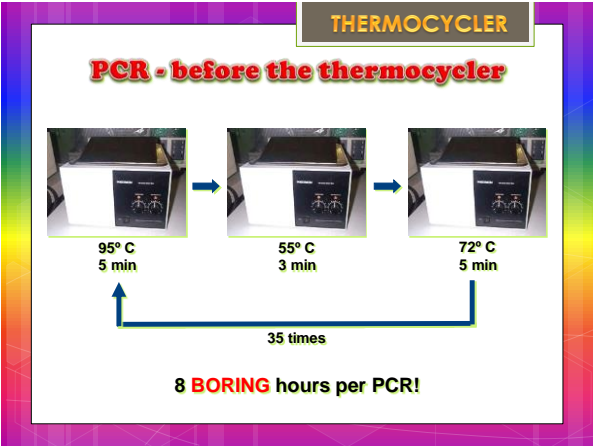
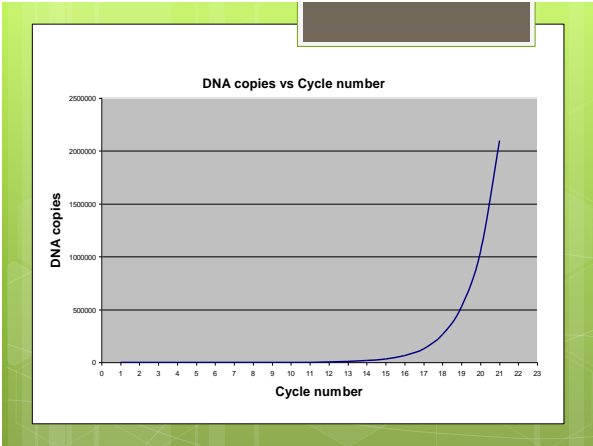


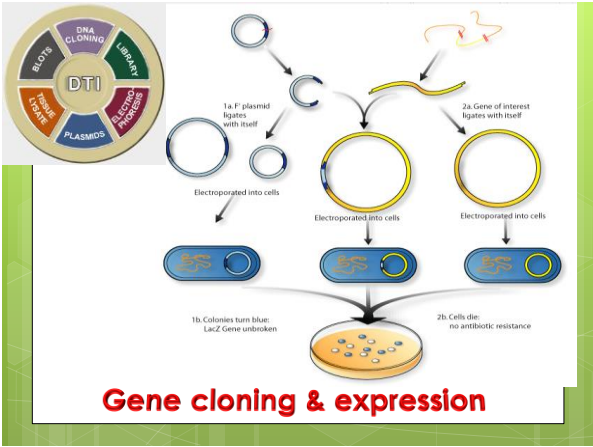
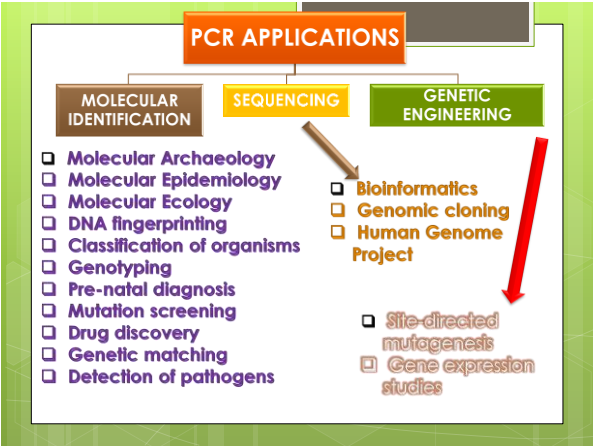
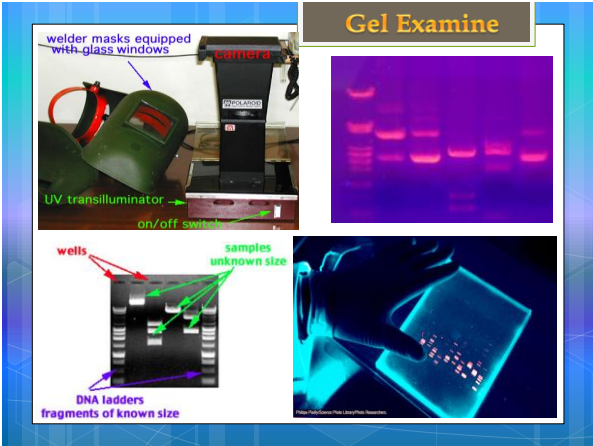
PCR tube



THERMOCYCLER







Creating mutation

Random or specific

ACCATCGGCCTGCATCA
TGGTAGCCTGACGTAGTCAT
GGTAGCCTGACT----

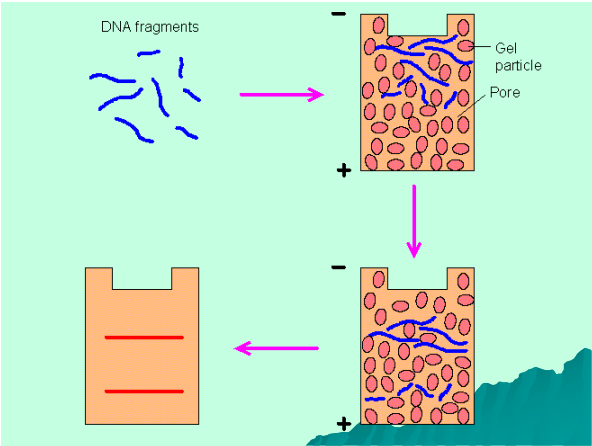


ELECTROPHORESIS:

Is the migration of charged molecules in solution in response to an electric field

The rate of migration depends on:

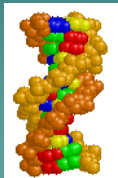
- ▣ Strength of the field
- ▣ Net charge
- ▣ Size & shape of the molecules
- ▣ Ionic strength, viscosity & temperature of the medium (in which molecules are moving)



AGAROSE GEL ELECTROPHORESIS

The *standard* method used to separate, identify & purify DNA fragments is **electrophoresis** through agarose gels

- The technique is
- ➡ Simple
 - ➡ Rapid to perform
 - ➡ Capable of resolving mixtures of DNA fragments
 - ➡ The location of DNA within the gel can be determined directly



(AGAROSE)

Is **extracted** from seaweed & is a linear polymer

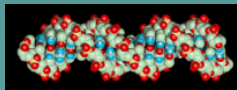


Commercially available agarose is not completely pure; it is contaminated with other polysaccharides, salts, & proteins

The amount of contamination varies from batch to batch & from manufacture to manufacture

These difference can affect on

- ➡ Migration of the DNA
- ➡ Ability of the DNA recovered from gel



Now, most manufacturers prepare **special grades** of agarose that are screened for the presence of inhibitors & nucleases & for minimal background fluorescence after staining with ethidium bromide

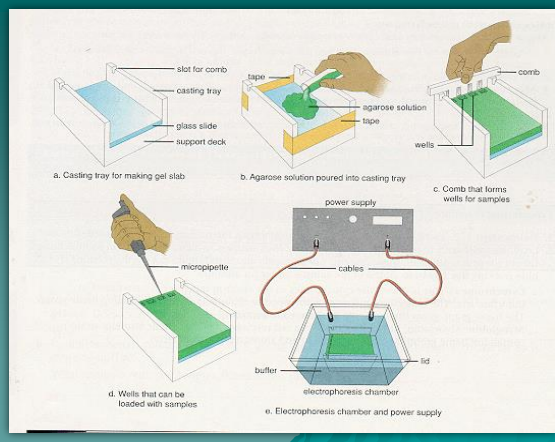
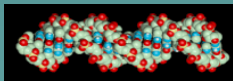


Agarose gels are cast by melting the agarose in the presence of the desired buffer until a clear, transparent solution is achieved

The melted solution is then poured into a mold & allow to harden.

The agarose forms matrix

When an electric field is applied across the gel, DNA which is negatively charged at neutral PH, migrates toward the **anode**



Agarose Concentration:

By using gels of different concentrations, it is possible to resolve a wide size of DNA molecules

Range of separation in gels containing different amounts of agarose

Amount of agarose in gel (%[w/v])	Efficient range of separation of linear DNA molecules (kb)
0.3	5-60
0.6	1-20
0.7	0.8-10
0.9	0.5-7
1.2	0.4-6
1.5	0.2-3
2.0	0.2-3

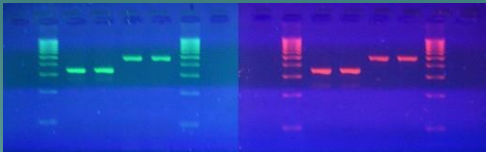
Presence of Intercalating Dyes:

- ✎ The central dye in agarose gel electrophoresis is *ethidium bromide*
- ✎ It has unique property of fluorescing under UV light when intercalated with DNA
- ✎ By running DNA through an EtBr-treated gel & exposing it to UV light, distinct bands of DNA become visible

Ethidium Bromide is a carcinogen & should be handled with care

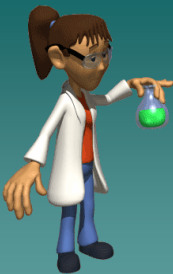
Other dyes are sometimes used including SYBER green or SYBER safe. SYBER dyes are thought to be less carcinogenic than EtBr & to give cleaner, higher powered staining

Midori Green DNA Stain



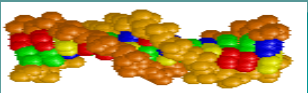
Left side Gel stained with Midori green stain & Right side Gel Stained with Ethidium Bromide

Electrophoresis Buffer:



Several different buffers are available for electrophoresis (TAE, TPE & TBE)

Electrophoresis buffers are usually made up as concentrated solutions & stored at room temp



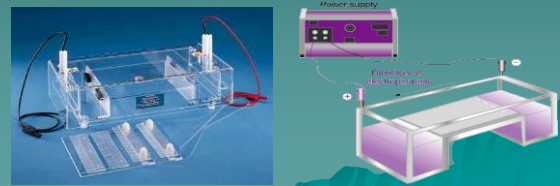
Commonly used Electrophoresis Buffers

Buffer	Working solution	Conc. Stock solution (Per Liter)
Tris-acetate (TAE)	1X: 0.04 M Tris-acetate 0.001 M EDTA	50X: 243 g Tris base 57.1 ml glacial acetic acid 100 ml 0.5M EDTA (PH 8.0)
Tris-borate (TBE)	0.5X: 0.045 M Tris-borate 0.001 M EDTA	5X: 54 g Tris base 27.5 g boric acid 20 ml 0.5M EDTA (PH 8.0)

Apparatuses Used For Agarose Gel Electrophoresis

The most commonly used configuration is the horizontal slab gel

Horizontal slab gels are usually poured on a glass plate or plastic tray that can be installed on a platform in electrophoresis tank



Preparation & Examination of Agarose gels

1. Seal the edges of a clean, dry, glass plate (plastic) with tape so as to form a mold. Set the mold on a horizontal section of bench

2. Prepare the materials you will need



3. Prepare mixture

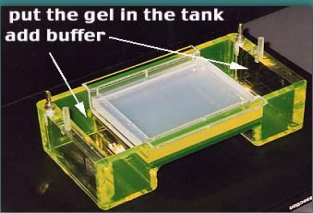


4. Gel the mixture

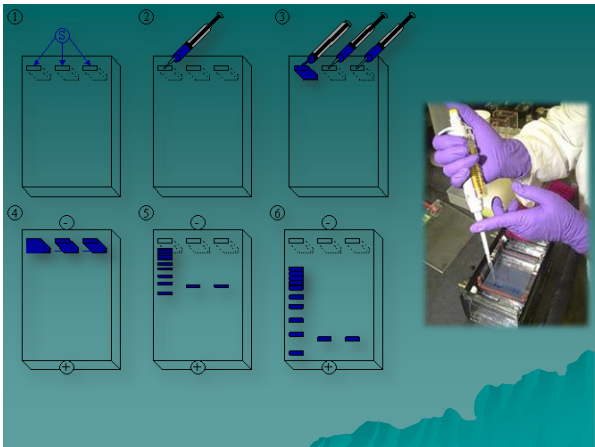
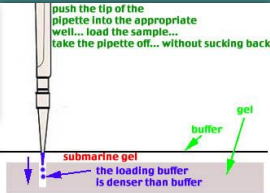
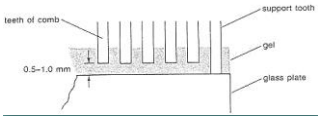
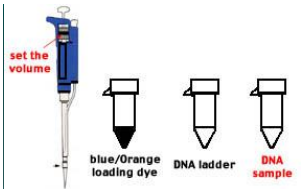
5. Cast gel (add ethidium bromide to the cooled gel)

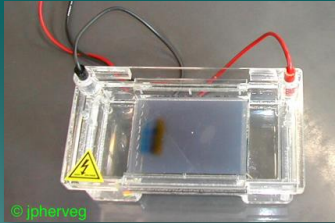


6. Place mold in the tank



7. Charge the wells

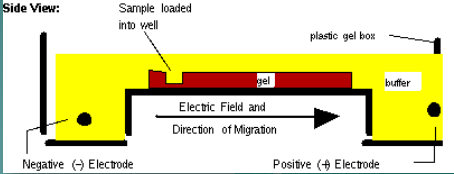





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8. Plug in

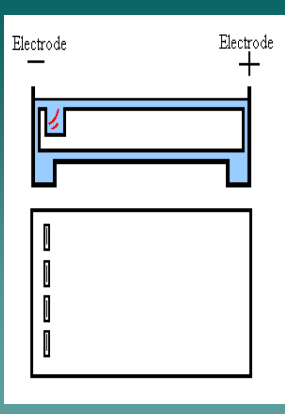
Side View:



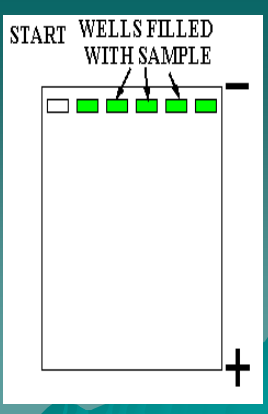


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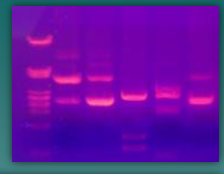


START WELLS FILLED WITH SAMPLE



9. Examine gel





UV transilluminator

